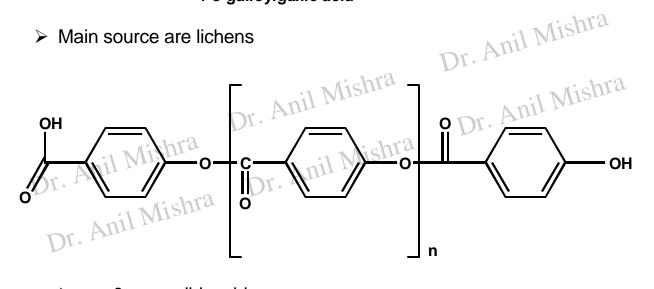
## Depsides

- These are compounds formed by the interaction of the carboxylic group of one phenolic acid molecule and the hydroxyl of the other.
- May be considered to be intermolecular esters formed from two or more molecules of the same or different phenolic benzoic acids. Depending on the number of the component units, di-, tri-, etc. depsides may result.

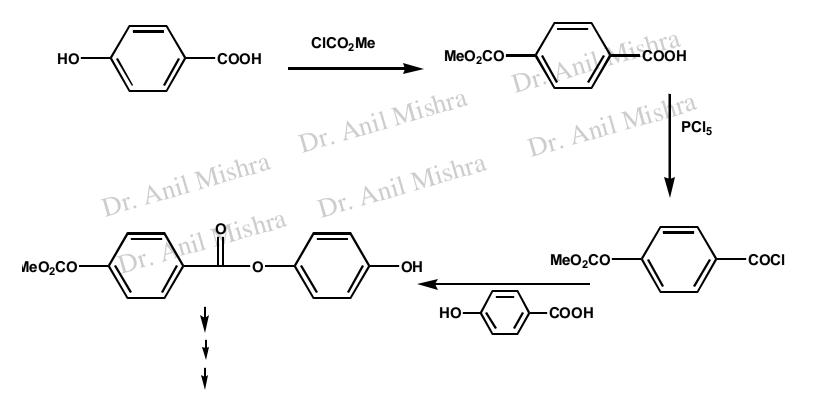


4-o-galloylgallic acid



| $\triangleright$ | n = 0 | didepside  |
|------------------|-------|------------|
| $\triangleright$ | n = 1 | tridepside |

## Synthesis



- By using different phenolic acids, a large variety of depsides can be synthesized
- When hydroxyl group is *meta* or *para* to the carboxylic group, the phenolic acid is readily carboxymethylated, but *ortho* groups are very resistant under the same conditions due to stearic effect.